

MAASAI MARA UNIVERSITY

REGULAR UNIVERSITY EXAMINATIONS 2017/2018 ACADEMIC YEAR FOURTH YEAR SECOND SEMESTER

SCHOOL OF SCIENCE BACHELOR OF SCIENCE CHEMISTRY

COURSE CODE: CHE 417 COURSE TITLE: BIOINORGANIC CHEMISTRY

DATE: 20TH APRIL 2018

TIME: 1100 - 1300 HRS

INSTRUCTIONS TO CANDIDATES

- 1. Answer Question **ONE** and any other **TWO** questions
- 2. No writing on the Question paper
- 3. Use of mobile phone in the exam room is prohibited

This paper consists of TWO printed pages. Please turn over.

Question one (30 marks)

(a)Define /explain the following *i*. Ferittin

(12 marks)

ii. Siderophores	
iii. Hemethrins	
iv. Metalloenzyme	
v. Cooperativity	
vi. Cytochrome	
(b) Explain the functions of hemoglobin	(6 marks)
(c) Explain the term lanthanide contraction	(2 marks)
(d) State the reason why Ce ³⁺ is a powerful oxidising agent and	l find
application as an oxidizing agent while Eu ²⁺ is a reducing ag	gent find
applications as a reducing	(4 marks)
(e) Giving at least two examples distinguish between hard and	
	(3 marks)
(f) State three similarities between lanthanides and actinides	(3 marks)
Question two (20 marks) (a)(i) With help of an illustration of a curve explain cooperativity	i.e. the
variations in affinity of myoglobin and oxygen to heamoglobin	
variations in affinity of myoglobin and oxygen to heamoglobin (5 m	arks)
variations in affinity of myoglobin and oxygen to heamoglobin	
variations in affinity of myoglobin and oxygen to heamoglobin (5 m) (ii) List five applications of Lanthanides	arks) (5 marks)
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- (a) Define the term photosynthesis and describe the internal and external factors that affect the rate of photosynthesis. (10 marks)
- (b) Explain transport, formation and degradation of hydrogen carbonate in our body (10 marks)

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